

Metabolism of the Prenylated Pterocarpan Edunol by *Aspergillus flavus*

Satoshi Tahara

Department of Agricultural Chemistry, Faculty of Agriculture, Hokkaido University, Kita-ku, Sapporo 060, Japan

John L. Ingham

Department of Food Science, Food Studies Building, University of Reading, Whiteknights, P.O. Box 226, Reading RG6 2AP, England

Z. Naturforsch. **42c**, 1050–1054 (1987); received June 10, 1986

Isoflavonoid, Prenylated pterocarpan, Edunol, Fungal metabolism, *Aspergillus flavus*

When incubated in liquid culture with *Aspergillus flavus*, the prenylated pterocarpan (–)-edunol [2-(3,3-dimethylallyl)-3-hydroxy-8,9-methylenedioxypterocarpan (**1**)] was converted into a dihydrofurano-pterocarpan (**2**), a dihydropyrano-pterocarpan (**3**), and a 2,3-dihydro-dihydroxyphenyl-substituted pterocarpan (**4**).

Introduction

In our previous papers, we reported that isoflavones with a 3,3-dimethylallyl (prenyl) substituent at C-6, C-8 (ring A) or C-3' (ring B) were variously metabolized by the fungus *Aspergillus flavus* to give hydrates (luteone [1] and wighteone [2]), and derivatives possessing dihydrofurano, dihydropyrano or 2,3-dihydrodihydroxyphenyl side-attachments (luteone [1], wighteone [2], 2,3-dehydrokievitone [3], licoisoflavone A [4], and 2'-hydroxylupalbigenin [5]). Studies involving the plant pathogenic fungus *Fusarium oxysporum* f. sp. *phaseoli* have shown that two other prenylated isoflavonoids, kievitone [5,7,2',4'-tetrahydroxy-8-(3,3-dimethylallyl)isoflavanone] and phaseollidin [3,9-dihydroxy-10-(3,3-dimethylallyl)pterocarpan] are also metabolized *in vitro* to give the corresponding hydrates [6, 7]. In kievitone and phaseollidin, the prenyl groups are located on different aromatic rings, and their hydration by *F. oxysporum* f. sp. *phaseoli* suggests that the enzyme responsible (kievitone hydratase [8]) is relatively non-specific in its action, or the fungus contains two hydratases differing in the substrate specificity [9].

In contrast, *A. flavus* exhibits a higher degree of substrate preference bringing about hydration of prenyl groups at C-6 on ring A of isoflavones (luteone and wighteone [1, 2], but not those located at C-8 (also ring A; 2,3-dehydrokievitone [3]) or C-3' (ring B; licoisoflavone A and 2'-hydroxylupalbigenin [4,

5]). The present study involving (–)-edunol [2-(3,3-dimethylallyl)-3-hydroxy-8,9-methylenedioxypterocarpan (**1**)], a fungitoxic [10] isoflavonoid from the root bark of *Neorautanenia edulis* (Leguminosae) [11], was undertaken to determine if *A. flavus* could: a) metabolize an isoflavonoid of a type different from the isoflavones previously tested, and b) convert the prenyl group at C-2 (\equiv C-6 of isoflavones) into a hydrated sidechain analogous with that encountered in the hydrates of luteone and wighteone [1, 2].

Results and Discussion

When shaken for 4 days (25 °C) in a liquid medium with *A. flavus*, (–)-edunol (**1**) was gradually metabolized to give three laevorotatory products designated ED-AF-1, ED-AF-2 (both M^+ 368), and ED-AF-3 (M^+ 386). These compounds, together with unchanged edunol (M^+ 352), were extracted from the medium with EtOAc, and then were separated by preparative Si gel TLC (PTLC) as outlined in the Experimental section. Their characterization as pterocarpan **2–4** is described in this report. Yields and comparative R_F values for each metabolite, and for the substrate, are shown in Table I. Surprisingly, no evidence was obtained to indicate that edunol could be converted to a hydrate derivative (expected M^+ 370) despite the fact that *A. flavus* has been found to readily hydrate the isoflavones luteone and wighteone [1, 2], each of which is prenylated at the position (C-6) equivalent to C-2 in **1**.

The UV (MeOH) spectrum of metabolite ED-AF-1 (M^+ 368; = substrate + [O]) was unaffected by NaOMe indicating the absence of a phenolic OH

Reprint requests to Dr. S. Tahara.

Verlag der Zeitschrift für Naturforschung, D-7400 Tübingen
0341–0382/87/0900–1050 \$ 01.30/0



Dieses Werk wurde im Jahr 2013 vom Verlag Zeitschrift für Naturforschung in Zusammenarbeit mit der Max-Planck-Gesellschaft zur Förderung der Wissenschaften e.V. digitalisiert und unter folgender Lizenz veröffentlicht: Creative Commons Namensnennung-Keine Bearbeitung 3.0 Deutschland Lizenz.

Zum 01.01.2015 ist eine Anpassung der Lizenzbedingungen (Entfall der Creative Commons Lizenzbedingung „Keine Bearbeitung“) beabsichtigt, um eine Nachnutzung auch im Rahmen zukünftiger wissenschaftlicher Nutzungsformen zu ermöglichen.

This work has been digitalized and published in 2013 by Verlag Zeitschrift für Naturforschung in cooperation with the Max Planck Society for the Advancement of Science under a Creative Commons Attribution-NoDerivs 3.0 Germany License.

On 01.01.2015 it is planned to change the License Conditions (the removal of the Creative Commons License condition “no derivative works”). This is to allow reuse in the area of future scientific usage.

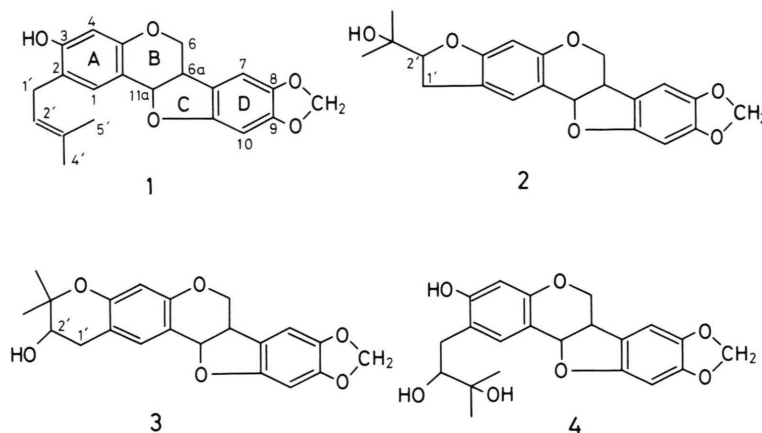


Table I. Chromatographic properties and yields of edunol and its metabolites.

Compound	R_F (TLC) value ^a		Yield ^b	
	CM (25:1)	CAAm (35:30:1)	[mg]	[%]
Edunol (1)	0.86	0.82	7.4	24.7
ED-AF-1 (2)	0.85	0.78	3.0	9.6
ED-AF-2 (3)	0.80	0.78	8.4	26.8
ED-AF-3 (4)	0.26	0.27	5.9	17.9

^a Solvent system abbrev., CM = CHCl_3 -MeOH; CAAm = CHCl_3 -acetone-conc. aqueous NH_3 .

^b Yield in mg from 30 mg of substrate (**1**); % yield is on a molar basis.

group (*cf.* the alkaline spectrum of **1** with a C-3 hydroxyl substituent). This observation, and the difference of 16 atomic mass units between the molecular weight of **1** and ED-AF-1, can be explained by the fungal-mediated formation of an ether ring side-attachment involving the substrate prenyl (C-2) and the *ortho* (C-3)-located OH group.

In the ^1H NMR spectrum of ED-AF-1, this side-attachment afforded a set of signals [δ 1.21 and 1.24 (both 3H, two s, 4'- and 5'- H_3), 3.07 and 3.26 (both 1H, two dd, $J = 16.3$ & 9.3 Hz, and 16.3 & 8.3 Hz, 1'- H_a and 1'- H_b), and 4.64 (1H, dd, $J = 9.8$ & 8.3 Hz, 2'-H)] with chemical shift values closely resembling those given by the 2-(1-hydroxy-1-methylethyl)-2,3-dihydrofuran substituent previously found in several isoflavones including luteone metabolite BC-1 [1] and lupinisoflavones B-F [12]. Prominent MS fragments at m/z 309 ($M^+ - 59$; 11%) and 59 (24%) were also consistent with the presence of such a side-

attachment [1]. ^1H NMR signals attributable to the aromatic (A/D) and heterocyclic (B/C) ring protons, and the $\text{O}-\text{CH}_2-\text{O}$ group, of edunol (Table II) were similarly evident in the spectrum of ED-AF-1, thereby permitting the metabolite to be formulated as shown in **2**. This structure has already been assigned to "neoplanol", a racemic dihydrofurano-pterocarp produced by chemical modification of edunol [13], but not as yet reported to be a natural product. MS and ^1H NMR data obtained for ED-AF-1 were in good agreement with those published [13] for synthetic "neoplanol" (see Table II, and the Experimental section for comparative details).

The ^1H NMR spectrum of ED-AF-2 ($M^+ 368$; = substrate + [O]) clearly revealed that this major, non-phenolic *Aspergillus* metabolite differed from **2** only with respect to the nature of the A-ring side-attachment (Table II). Thus, instead of a dihydrofurano substituent, a set of aliphatic proton signals at δ 1.25 and 1.33 (both 3H, two s, 4'- and 5'- H_3), 2.62 and 3.01 (both 1H, two dd, $J = 16.5$ & 7.3 Hz, and 16.5 & 5.4 Hz, 1'- H_a and 1'- H_b) and 3.79 (1H, ddd, $J = 7.3$, 5.4 & 5.1 Hz, 2'-H coupled to 2'-OH with $J = 5.1$ Hz) defined the presence of a 2,3-dihydro-3-hydroxy-2,2-dimethylpyrano side-attachment as in luteone metabolite BC-2 [1], 2,3-dehydrokievitone metabolite DK-M2 [3], and licoisoflavone A metabolites M-1-1 and M-3-1 [4]. As with these four metabolites, the MS of ED-AF-2 afforded a characteristic fragment at $M^+ - 71$ (m/z 297; 98%) [1, 3, 4]. Metabolite ED-AF-2 must therefore have structure **3**. Neorautanol from *Neorautanenia amboensis* has also been formulated as **3**. MS and ^1H NMR data

Table II. ^1H NMR data (δ values) for edunol and its *Aspergillus flavus* metabolites^a

Compound Proton	Edunol (1)	ED-AF-1 (2) ^b	ED-AF-2 (3) ^b	ED-AF-3 (4)
1-H	7.16 s	7.22 s	7.15 s	7.22 s
4-H	6.38 s	6.21 s	6.23 s	6.34 s
7-H	6.88 s	6.89 s	6.90 s	6.89 s
10-H	6.40 s	6.40 s	6.39 s	6.40 s
6-H _{eq}	4.24 m	4.27 m	4.26 m	4.27 m
6-H _{ax}	3.55 m	3.58 m	3.60 m	3.57 m
6a-H	(2H)	(2H)	(2H)	(2H)
11a-H	5.46 br. d $J=6.4$	5.50 br. d $J=6.4$	5.49 br. d $J=6.1$	5.47 br. d $J=6.1$
O-CH ₂ -O	5.90 d $J=1.0$	5.91 d $J=1.0$	5.91 d $J=1.0$	5.91 d $J=1.0$
	5.93 d $J=1.0$	5.94 d $J=1.0$	5.94 d $J=1.0$	5.93 d $J=1.0$
1'-H _a	3.29 br. d (2H) $J=7.3$	3.07 dd $J=16.3, 9.3$	2.69 dd $J=16.6, 7.3$	2.62 dd $J=14.5, 9.5$
1'-H _b		3.26 dd $J=16.3, 8.3$	3.01 dd $J=16.6, 5.4$	2.93 dd $J=14.5, 2.3$
2'-H	5.35 br. t $J=7.3$	4.64 dd $J=9.3, 8.3$	3.79 ddd $J=7.3, 5.4, 5.1$	3.64 dd $J=9.4, 2.3$
4'-CH ₃	1.73 s (3H)	1.21 s (3H)	1.25 s (3H)	1.25 s (3H)
5'-CH ₃	1.74 s (3H)	1.24 s (3H)	1.33 s (3H)	1.27 s (3H)
2'-OH	—	—	4.23 d $J=5.1$	—
3'-OH	—	3.62 s	—	—

^a All spectra were determined in acetone- d_6 at 100 MHz (TMS reference). Coupling constants (J) are in Hz. For multiplets, the δ value indicates the centre of the signal.

^b ^1H NMR chemical shift values for synthetic "neoplanol" (= **2**) and natural neorautanol (= **3**) in CDCl_3 (80 MHz) are reported in ref. [13].

obtained for ED-AF-2 agreed reasonably with those of neorautanol [13].

The third metabolite, ED-AF-3 (M^+ 386; = substrate + $2 \times [\text{OH}]$), ran well below **2** and **3** on thin-layer chromatograms developed in both CM and CAAM (Table I). This increased polarity relative to the other *Aspergillus* metabolites, coupled with the detection of an intense MS ion at $M^+ - 89$ (m/z 297; 100%), suggested that ED-AF-3 contained a 2,3-dihydrodihydroxyphenyl side-chain [1, 3, 4]. An underivatized C-3 OH group was apparent from the UV (MeOH + NaOMe) maximum at 251 nm [14].

Structure **4** for ED-AF-3 was confirmed by the ^1H NMR spectrum which exhibited signals typical of a 1,2-glycol type side-chain at δ 1.25 and 1.27 (both 3H, two s, 4'- and 5'-H₃), 2.62 and 2.93 (both 1H, two dd, $J=14.5$ & 9.4 Hz, and 14.5 & 2.3 Hz, 1'-H_a and 1'-H_b) and 3.64 (1H, dd, $J=9.4$ & 2.3 Hz, 2'-H). In all other respects, the ^1H NMR spectrum closely resembled that of edunol (Table II). Apart from its appearance in ED-AF-3, the 2,3-dihydrodihydroxyphenyl (2,3-dihydroxy-3-methylbutyl) side-chain has also been encountered in luteone glycol [1], wight-eone glycol [2], 2,3-dehydrokievitone glycol [3] and

licoisoflavone A glycol [4], all of which are produced by *A. flavus* from the corresponding isoflavone substrates.

Experimental

General procedures (e.g. silica gel PTLC, and m.p., UV, MS and ^1H NMR measurements) were undertaken using the equipment and conditions previously described [4, 12]. The substrate pterocarpan (**1**) was isolated from the shredded root bark of *Neorautanenia edulis* [11].

Edunol (**1**) [2-(3,3-dimethylallyl)-3-hydroxy-8,9-methylenedioxypterocarpan]

Colourless needles, m.p. 151–152 °C; $[\alpha]_{\text{D}}^{23} - 325^\circ$ ($c = 0.088$, MeOH). UV: λ_{max} , nm: MeOH 232 sh, 288 sh, 294, 310; + NaOMe, 251, 304. MS (rel. int. %): m/z 353 ($\text{M}^+ + 1$; 25), 352 (M^+ ; 100), 298 (14), 297 ($\text{M}^+ - 55$; 71), 296 (31), 176 (12), 175 (21), 148 (16), 147 (13), 71 (12), 69 (12), 57 (21).

Metabolic experiments

Aspergillus flavus (isolate AHU 7049) was cultured for 4 days in a shaking liquid medium consisting of glucose (5 g), peptone (1 g), yeast extract (0.1 g) and H_2O (100 ml). A solution of **1** (5 mg in 1 ml of EtOH) was then added, and after a further 4 days incubation the metabolites, and any remaining substrate, were extracted from the medium with EtOAc (see ref. [1] for exact details).

Isolation and purification of edunol metabolites

An EtOAc extract of the combined *Aspergillus* culture medium (600 ml, initially containing 30 mg of **1**) from 6 flasks was washed with 5% aqueous NaHCO_3 , and then with a saturated solution of NaCl. After removal of the EtOAc *in vacuo* (40 °C), the residue was chromatographed (Si gel preparative TLC) in CHCl_3 -acetone-conc. aqueous NH_3 (CAAm; 35:30:1) to afford unchanged edunol (upper band), a mixture of ED-AF-1 and ED-AF-2 (middle band), and ED-AF-3 (lower band). See Table I for R_F values. After elution with EtOAc, edunol and ED-AF-3 were rechromatographed in CHCl_3 -MeOH (CM; 25:1) to give the pure pterocar-

pans. After concentration, the eluate of the middle band (CAAm chromatogram) deposited colourless plates of ED-AF-2. The mother liquor was then chromatographed (Si gel PTLC in CM, $\times 3$) to afford ED-AF-1 (upper zone) and a further quantity of ED-AF-2 (lower zone). The yields of each metabolite are given in Table I.

Metabolite ED-AF-1 (**2**; = "neopranol" [13])

Colourless fine rods, m.p. 204–205 °C (subliming at $>200^\circ\text{C}$); $[\alpha]_{\text{D}}^{23} - 340^\circ$ ($c = 0.096$, MeOH). UV: λ_{max} , nm: MeOH 232 sh, 291 sh, 296 sh, 302, 308 sh; + NaOMe, no change. MS (rel. int. %): m/z 369 ($\text{M}^+ + 1$; 24), 368 (M^+ ; 100), 335 (16), 310 (18), 309 ($\text{M}^+ - 59$, 11), 297 (27), 296 (12), 175 (18), 162 (25), 160 (13), 147 (13), 71 (16), 59 (24), 57 (29). MS data published for synthetic "neopranol" [13] are: m/z 368 (M^+ ; 100), 353 (2), 335 (14), 310 (16), 297 (25), 175 (19), 162 (29), 160 (10), 151 (12), 148 (15).

Metabolite ED-AF-2 (**3**; = neorautanol [13])

Colourless plates, m.p. 207–209 °C (subliming at $>195^\circ\text{C}$; neorautanol [13], m.p. 93–95 °C); $[\alpha]_{\text{D}}^{23} - 271^\circ$ ($c = 0.136$, MeOH). UV: λ_{max} , nm: MeOH 232 sh, 285 sh, 289 sh, 295, 310; + NaOMe, no change. MS (rel. int. %): m/z 369 ($\text{M}^+ + 1$; 24), 368 (M^+ ; 100), 335 (8), 298 (22), 297 ($\text{M}^+ - 71$; 98), 296 (27), 267 (9), 175 (9), 162 (17), 85 (8), 71 (16), 69 (12), 57 (27).

Metabolite ED-AF-3 (**4**)

Colourless fine rods, m.p. 195–197 °C (subliming at $>185^\circ\text{C}$); $[\alpha]_{\text{D}}^{23} - 277^\circ$ ($c = 0.188$, MeOH). UV: λ_{max} , nm: MeOH 227 sh, 281 sh, 287 sh, 291.5, 310.5; + NaOMe, 251, 303.5. MS (rel. int. %): m/z 387 ($\text{M}^+ + 1$; 22), 386 (M^+ ; 92), 368 ($\text{M}^+ - 18$; 34), 298 (23), 297 ($\text{M}^+ - 89$; 100), 296 (20), 175 (11), 162 (15), 151 (14), 85 (11), 71 (21), 69 (16), 59 (13), 57 (38).

Acknowledgements

We thank Professor S. Takao for kindly supplying the *Aspergillus* culture, and Miss S. Endo (^1H NMR) and Mr. K. Watanabe and Miss Y. Atsuta (MS) for spectroscopic measurements. Financial support (to S. T.) by a grant for scientific research (No. 61560130) from the Ministry of Education, Science and Culture of Japan is also gratefully acknowledged.

- [1] S. Tahara, S. Nakahara, J. Mizutani, and J. L. Ingham, *Agric. Biol. Chem.* **48**, 1471 (1984).
- [2] S. Tahara, S. Nakahara, J. L. Ingham, and J. Mizutani, *Nippon Nōgeikagaku Kaishi* **59**, 1039 (1985).
- [3] S. Tahara, E. Misumi, J. Mizutani, and J. L. Ingham, *Z. Naturforsch.* **42c**, 1055–1062 (1987).
- [4] S. Tahara, S. Nakahara, J. Mizutani, and J. L. Ingham, *Agric. Biol. Chem.* **49**, 2605 (1985).
- [5] S. Nakahara, S. Tahara, J. Mizutani, and J. L. Ingham, *Agric. Biol. Chem.* **50**, 863 (1986).
- [6] P. J. Kuhn, D. A. Smith, and D. F. Ewing, *Phytochemistry* **16**, 296 (1977).
- [7] D. A. Smith, P. J. Kuhn, J. A. Bailey, and R. S. Burden, *Phytochemistry* **19**, 1673 (1980).
- [8] P. J. Kuhn and D. A. Smith, *Physiol. Plant Path.* **14**, 179 (1979).
- [9] H. D. Vanetten, D. E. Matthews, and D. A. Smith, *Phytoalexins* (J. A. Bailey and J. W. Mansfield, eds.), p. 192, Blackie & Son Ltd., Glasgow 1982.
- [10] J. L. Ingham, unpublished results.
- [11] G. J. H. Rall, J. P. Engelbrecht, and A. J. Brink, *J. South Afr. Chem. Inst.* **24**, 56 (1971).
- [12] S. Tahara, J. L. Ingham, S. Nakahara, J. Mizutani, and J. B. Harborne, *Phytochemistry* **23**, 1889 (1984).
- [13] J. C. Breytenbach and G. J. H. Rall, *J. Chem. Soc. Perkin Trans. I* **1980**, 1804.
- [14] J. L. Ingham and K. R. Markham, *Phytochemistry* **19**, 1203 (1980).